Determination of Moisture in Powder and Lyophilised Saffron (*Crocus sativus* L.) by Karl Fischer Method

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Abstract: Saffron is a spice obtained from the dried stigmas of the purple-flowered saffron crocus (*Crocus sativus* L.). The determination of the water content in saffron is very important in order to establish its microbiological stability and consequently to determine the commercial value. The average moisture content in dried saffron powder ranges between 10-12% causing a slow degradation of the product. On the other hand lyophilised saffron has lower water content and a better stability. The reference method commonly used for the determination of moisture in saffron it is based on the gravimetric loss of weight after heating at 105 °C for 16 hours.

In this paper a procedure to obtain lyophilised saffron is proposed and two different methods, namely the infrared balance and the Karl Fischer titration, are used to determine moisture in dried and lyophilised saffron powder, and results are compared with the reference method. The Karl Fischer method, never applied so far to determine moisture in saffron, gave the best results. The average values determined were 6.87% and 4.01% (w/w) for conventionally dried and lyophilised saffron powder, respectively. The Karl Fischer titration method allows to determine in a faster manner the moisture in saffron powder and in lyophilised saffron, without any interference and using a minor quantity of saffron compared to the gravimetric procedure reported in the reference method. The volatile matter content, differently from the reference method, this way is not determined.

Keywords: Water, moisture, Karl Fischer, lyophilised saffron, saffron powder, crocins.

INTRODUCTION

Saffron is a spice obtained from the dried yellowish orange 2.5-4 cm long stigmas of the purple-flowered saffron crocus (Crocus sativus L.) of the iris family, Iridaceae, native to Southern Europe and the Middle East, and it is so named from Arabic term "zaferan", meaning yellow. Saffron has an aromatic hot and slightly bitter taste and contains different water-soluble colouring substances, the crocins, which provide a golden-coloured dye effective even at a dilution of 1:100.000 (w/v). This spice has been used since ancient times as a food additive and as dye, and it has been also used as a drug to treat various human health conditions [1]. Recently, its properties as antidepressant [2] and as antitumoral against different malignant cells have been demonstrated [3, 4]. Saffron is among the most valuable, most expensive spices in the world, since harvesting saffron is very labourintensive, considering the low number of flowers per hectar and the work required for harvesting it [5]. Each blossom has three stigmas, which are picked by hand in the morning before the heat of the day, and then dried by heating or sun dried for 15-30 minutes, while the blossoms are thrown away. Saffron is also very hygroscopic, the exposure to moisture creating a risk of spoilage of the product. It has been determined for a typical saffron sample, before drying, that the relative humidity is about 70%, while the essential oil content in particular of the picrocrocin (saffron-bitter) is between 0.4-1.3% (w/w), and the water content is 9-10% (dried final product) and 12-18% (wooden packaging) [6]. The postharvest treatment of drying stigmas is necessary to reduce the moisture to a value below 15% and to obtain a product microbiologically stable in time. To the date, drying is carried out at temperatures comprised between 20 °C and 120 °C, depending on the local tradition and on the procedure used, usually until the colour of the stigmas changes from orange-red to dark red, as an evidence of the end of the drying. The practice of sun drying stigmas (prevalent in India) was perceived likely as a major cause of poor quality and alternative methods were studied in order to determine their effectiveness and commercial practicability [7, 8]. In sun drying, stigmas are dried for 3-5 days in the sun till the moisture content is reduced to 10-12%. Alternatively, during flowers harvest, the whole flowers are dried in the sun and stigmas are picked by hand from the dried flowers later [9]. The dried, warm saffron is then allowed to cool in a dry place [10]. Exposing the fresh stigmata of saffron to a hot source for many hours can cause reactions to happen between the native components of saffron like crocins and water, part of which evaporates.

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The crocins are the glycosyl esters of crocetin $(C_{44}H_{64}O_{24})$ that are unusually water-soluble carotenoids responsible for the strong colouring capacity of saffron. Picrocrocin $(C_{16}H_{26}O_7)$ a monoterpene glycoside and safranal $(C_{10}H_{14}O)$ are the principal bitter tasting substances in saffron. Safranal, a monoterpene aldehyde, is formed in saffron by hydrolysis during drying and storage from picrocrocin and is responsible for its aroma [11]. The water content in the dried product directly influences the amount of these valuable compounds contained in the spice.

Recently, different papers about the optimum drying conditions to obtain the best hot dried spice [12] and about dehydration postharvesting treatment to reduce moisture in stigmas were published [7, 8].

In the literature different aspects of saffron, principally referring to the analysis and identification of main components have been evaluated, as well as the identification of bioactive molecules contained in saffron like crocins for their possible use in medicine [3] while no method has been proposed for the determination of water content in dried saffron powder except for the "determination of moisture and volatile matter" indicated at point 9 of International Standard Regulation ISO 3632-2-93, based on the gravimetric measure of the loss of weight in oven at 103 °C \pm 2 °C. According to the ISO, the determination gives the total of water and volatile matter present, without the possibility to separate the two amounts [13].

In this paper the Karl Fischer (KF) titration method [14] is proposed to determine the water content in saffron powder and in lyophilised saffron. The use of lyophilization procedure here described is new for saffron; it uses low temperatures in a high vacuum environment to dry the spice reducing the risk of reactions between the native substances contained in the spice and allowing to obtain saffron with a constant content of the native compounds. The results obtained with the KF and gravimetric infrared balance methods are compared to those obtained with the official ISO method.

MATERIALS AND METHODOLOGY

Materials

Saffron was obtained from two Italian producers, namely Sanna, S. Gavino Monreale (Cagliari) and Navelli, Civitaretenga (L'Aquila), and one from an Iranian producer, Bahraman Saffron (Masshad). All saffron powder samples were classified according to the ISO classification as Category I (see Table 1). Lyophilised saffron was prepared as described below. For each of the four saffron typology under study (Sanna, Navelly, Bahraman, lyophilised) five aliquots were sampled and analysed using the ISO, the infrared balance and KF method, respectively, to determine the water content. Each sample was analysed in triplicate and results were reported as the average. Considering that for the saffron powder the analytical identification and hence the commercial classification according to the ISO 3632-2-93 [13] is based on the spectrophotometric determination of crocins, picrocrocins and safranal since their amount in the dried product is dependent on the moisture content. Crocins, picrocrocins and safranal have been identified and quantified using UV spectroscopy determination of the absorbance maxima at 257, 330 and 440 nm, for picocrocins, safranal and crocins, respectively.

Lyophilised Saffron Preparation

Lyophilised saffron was obtained from the sample furnished by the producer Sanna as following: 250 g of fresh stigmas of *Crocus sativus* L. were, immediately after collecting in field, frozen at -20 °C for twenty four hours; then they were lyophilised for other 24 hours, powdered by grinding to obtain 50 g of orange coloured powered saffron. Fig. (1) shows the aspect of the starting material used.

Determination of Moisture by Standard Gravimetric Method

2.5 g of saffron stigmas previously powdered by grinding, were dried in an oven (Intercontinental s.r.l.) at 103 °C \pm 2°C for 16 hours at atmospheric pressure; after cooling in the dryer, the sample was weighted to the nearest 0.0001 g. A test for constancy of mass was performed by additional drying steps of 1 h until the difference in the mass did not exceed 0.5 mg. Each sample was analyzed in triplicate and the final result reported as the average on all the determinations (ISO 3632-2-93).

Determination of Moisture Using the Infrared Balance

5.0 g of each sample of saffron powder were dried on an infrared balance (Eurotherm, Gibertini, Novate, Italy) at 105 °C until the weight was constant (the difference between two consecutive determination did not exceed 4.0 mg) for 5 minutes. Each measure required about 20 minutes. Each sample was analyzed in triplicate and the final result taken as the arithmetic average on all the determinations.

Determination of Moisture by Volumetric Karl-Fischer Titration

The determinations of moisture were carried out using a micro ST 2038 apparatus for Karl Fischer titration (Crison

Table 1.	Commercial Classification of Saffron According to the ISO 3632-2-93; E ^{1%}	a is the Specific Extinction Coefficient
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Category	"Bitterness" E ^{1%} ₂₅₇	"Odour" E ^{1%} 330	"Colour" E ^{1%} 440
I	≥70	20 -50	≥ 190
П	≥ 55	20 - 50	≥ 150
III	≥ 40	20 - 50	≥110
IV	≥ 30	20 - 50	≥80
Powder saffron (this paper)	100	35	240
Lyophilised saffron (this paper)	230	100	330

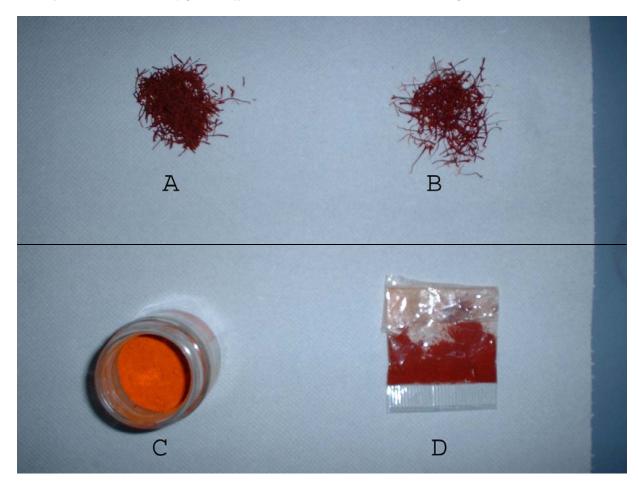


Fig. (1). Samples of iranian saffron (A), Sanna italian saffron (B), lyophilized saffron (C), and Navelli powder saffron (D).

Instruments SA, Spain). 70.0 mg of saffron lyophilised and 20.0 mg of powder saffron were analysed with Hydranal composite 5 (Hydranal solvent, Fluka) as reagent and methanol as solvent. The solvent was changed in each experiment. The analyses were performed in triplicate and the results averaged.

UV-VIS Spectrophotometric Analysis

Samples of saffron powder and lyophilised saffron were dissolved in water at a 0.04 % and 0.02 % (w/v), respectively and filtered through 0.20 μ m membrane (Millipore, Bedford, MA, USA) in order to eliminate rough impurities. They were then analysed using a spectrophotometer UV-VIS mod. UV-1601 (Shimadzu Corp., Tokyo, Japan) in the 200 – 600 nm wavelength range.

RESULTS AND DISCUSSION

Absorbance values limits used for the commercial classification of the powder saffron in the four existing categories [13] are reported in Table 1 for reference together with the values determined for the samples analysed in this study.

Table 2 shows the results obtained using the three methods above described for the water content determination using powder and lyophilised saffron.

Standard deviation is good for all of the three methods, but the values of moisture determined by oven drying (ISO method) are always higher than the others. This reflects the loss of weight caused by the chosen conditions and do not necessarily reflect the true water content: in fact the weight loss could be caused not only by the water, but also by the loss of volatile substances present in the powder or formed by degradation reactions related to the sample heating. The process of drying using the infrared balance is based on the irradiation of the sample with infrared radiation that is capable to induce the evaporation of the water contained in the sample. About 5 g of sample are required and the measure is relatively fast, about 20 minutes, while for the standard ISO method about 2.5 g and 16 hours are required.

In the literature many papers describe the use of the Karl Fischer titration method in the determination of the water content in foodstuffs. Recently its use in the determination of the water content in powdered milk was proposed [15, 16].

The Karl-Fischer titration [14, 17, 18] on the other hand is the only selective method even if, due to the nature of the titration, it is impossible to measure the moisture in a sample if the water content is too high. This method in fact is capable to determine very low moisture content in the sample and has the smallest standard deviation when applied to saffron (see Table 2). The quantity needed for the determination is very small, 70 mg and 20 mg for the dried and lyophilised saffron powder, respectively, and the determination is very fast, requiring about 1 minute.

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- Table 2.Weight Loss Percentage (w/w %) in: (A) Lyophilized Saffron, (B) Powder Saffron Sanna, (C) Powder Saffron Navelli, and
(D) Powder Saffron Iran, Determined by ISO Method, Infrared Balance and KF Titration. For the KF Method the Reported Values Coincide with the Water Content
- (**A**)

Lyophilized Saffron	Oven Drying (w/w %)	Infrared Balance (w/w %)	KF Titration (w/w %)
1	10.00	6.88	3.83
2	9.64	6.41	4.34
3	9.15	7.01	4.10
4	10.10	7.20	3.95
5	9.87	6.69	4.05
average	9.75	6.84	4.05
SD	0.31	0.25	0.16

(B)

Powder Saffron (Sanna)	Oven Drying (w/w %)	Infrared Balance (w/w %)	KF Titration (w/w %)
1	11.0	7.88	6.92
2	10.64	7.41	6.75
3	9.99	8.99	7.03
4	10.1	8.2	6.82
5	12.87	7.0	6.83
average	10.92	7.90	6.87
SD	0.95	0.62	0.09

(**C**)

Powder Saffron (Navelli)	Oven Drying (w/w %)	Infrared Balance (w/w %)	KF Titration (w/w %)
1	11.22	7.76	6.63
2	10.04	7.81	6.55
3	9.89	8.00	7.01
4	10.33	8.23	6.64
5	12.87	7.56	6.97
average	10.87	7.87	6.76
SD	1.01	0.21	0.17

(D)

Powder Saffron (Iran)	Oven Drying (w/w %)	Infrared Balance (w/w %)	KF Titration (w/w %)
1	11.22	7.36	6.83
2	10.04	7.21	6.95
3	9.08	8.04	7.10
4	10.53	8.13	6.34
5	12.97	7.16	7.05
average	10.77	7.58	6.85
SD	1.19	0.38	0.25

The determination of the water content of a sample requires a two step chemical reaction scheme that uses the reaction between iodine and water in presence of sulphur dioxide. The chemical reactions are the following:

(1)
$$CH_3OH + SO_2 + RN = [RNH]SO_3CH_3$$

(2) $[RNH]SO_{3}CH_{3} + I_{2} + H_{2}O + 2RN = [RNH]SO_{3}CH_{3} + 2[RNH]I$

The parameter measured in the volumetric KF method is the amount of iodine reduced in the second reaction step where [RNH]SO₃CH₃ act as catalyst. The data reported in the Table **2** indicate that lyophilised saffron (Sanna producer) has the lowest content of water, 4.01%, while the saffron conventionally dried obtained from the same producer has a value of 6.87%. Comparing the data reported in the Table **2** it can be observed that the official ISO method gives always an higher water content compared with the infrared balance and Karl Fischer titration method. For the KF method the values reported in Table **2** coincide with the water content, since any other substance present in the sample in very small amount should not interfere with the Iodine used in the titration.

To determine the content of glycosilated compounds present in the dried and lyophilised saffron, UV spectroscopy was used. The results show that lyophilised saffron water solution at 0.02 % (w/v) concentration has an higher values for the absorbance at 257, 330 and 440 nm, see Fig. (2A), compared to the saffron powder at a 0.04 % (w/v) concentration, see Fig. (2B). The higher absorbance at 440 nm is related to the highest crocins content, responsible for saffron colour, while the absorbance at 257 nm is related to the picrocrocins content, responsible for the taste, and the absorbance measured at 330 nm is related to the safranal content, responsible for the odour [19]. The higher crocins content in lyophilised saffron is due to its lower moisture content, in fact the moisture can catalyse the degradation (hydrolysis) of crocin into crocetin, causing also the fading of the colour in the saffron powder, the colour being inversely related to moisture content [9]. Liophylised saffron showed a colouring power ("Colour" $E_{440}^{1\%}$) of 330 according to the data reported in Fig. (2A) while powder saffron showed a value of 240. The value for the lyophilized saffron is way higher then the 190 considered acceptable for the Category I classification (see Table 1) indicating that this formulate can be classified in a new higher Category.

Moreover since the traditional drying method of the saffron is realized using high temperatures for a long time, the native compounds, e.g. the crocins, can be degraded, also explaining why the traditionally dried saffron has a lower content of these compounds compared with the lyophilised saffron powder, that has also a lower water content.

CONCLUSION

The obtained results using the official ISO, the infrared balance and the KF titration method to determine the water content in saffron powder and lyophilised saffron powder allowed us to conclude that the KF method could be an alternative technique to measure the water content in saffron with respect to the method suggested by the ISO 3632-2-93 [13] and ISO 5725-1 [20] regulations, based on the loss of weight after oven drying. The results obtained by KF titra-

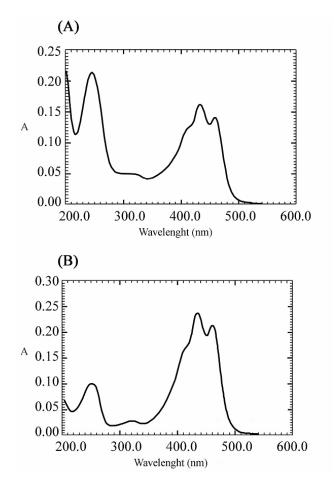


Fig. (2). UV-VIS (200–600 nm) spectra of lyophilised (A) and oven dried saffron (B); analysed solutions are at a 0.02 and 0.04 % (w/v) concentration, respectively.

tion allowed us to obtain a good repeatability, good accuracy considering that any volatile compound is not determined, and a faster determination. Moreover a smaller quantity of sample for the analysis compared to the oven drying method is required.

Lyophilized saffron is a new product, with an higher content of crocins, a constant content of native compounds, because of its minimum water content (about 4%) with respect to the traditional dried saffron and for this reasons it could be used as a standard substance to evaluate the quality and water content of saffron powder itself. At the same time we can conclude that the innovative method of lyophilisation here described, based on drying by freezing in a high vacuum, allows to obtain saffron with very low moisture content and consequently an higher crocins content, a longer shelf-life, a greater stability and an higher colouring power.

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