## **Open Access**

# A Novel Fructan Possessing DB Value from Roots of Arctium lappa L.

Shi Lei\*

College of Life Science, Qufu Normal University, Qufu, Shandong, 273165, China

**Abstract:** A fructan named ALF was isolated from the roots of *Arctium lappa* L.. The structure of ALF was identified by analyses of methylation, GC-MS, and both <sup>1</sup>H and <sup>13</sup>C NMR spectroscopy. The results obtained indicated that ALF, comprised of D-fructose and D-glucose in the molar ratio of 14:1, was an inulin-type fructan, which was confirmed by the composition of 14 fructose residues linked by  $\beta$  (2 $\rightarrow$ 1) glycosidic bond and 1 glucose residue linked by  $\alpha$  (1 $\rightarrow$ 2) glycosidic bond at the end of linear straight sugar chain.

Keywords: Fructan, Structure, DB (degree of branch) value, Arctium lappa L.

## INTRODUCTION

Fructans exist as a wide range of oligo- and polysaccharides in many species of bacteria, fungi, and plants [1]. They are classified into different families on the basis of their glycosidic linkages, consisting of  $(2\rightarrow 1)$ -linked  $\beta$ -Dfructofuranosyl units such as inulin, or  $(2\rightarrow 6)$ -linked  $\beta$ -Dfructofuranosyl units such as levans, or highly branched structures comprised of both  $(2\rightarrow 1)$ - and  $(2\rightarrow 6)$ -linked  $\beta$ -Dfructofuranosyl units such as graminans [2, 3].

Arctium lappa L., a fructan-containing member of the *Compositae* family, is a very popular edible vegetable in the orient countries. It has been extensively analyzed for its reserve and cell-wall polysaccharides [4], components having antimicrobial activity as well as for extractive components with antioxidant activity [5]. Kardosová A reported that a water-soluble fructan isolated from *Arctium lappa* L. has antitussive activity [6]. The present work reports on isolation, structural analysis of a fructan from the roots of *A. lappa* L., widely distributed in our geographical conditions.

#### MATERIALS AND METHODOLOGY

#### **Materials and Instruments**

T-series Dextran, DEAE-Sephadex A-50, and DEAEcellulose were purchased from Amersham Biosciences (Uppsala, Sweden). Trifluoroacetic acid (TFA) was purchased from Sigma Chemical Co. (St. Louis, MO, USA). All other chemical reagents were of analytical reagent grade.

NMR spectroscopy: AVANCE-600 superconductive NMR spectrometer (Bruker, Switzerland).

Optical rotation: WZZ-1S automatic polarimeter (Shanghai Physical Optics Instruments Co.).

IR: Perkin-Elmer 591B spectrophotometer with a KBr pellet (native ALF) or Nujol film (permethylated ALF).

GLC: Shimadzu-14B apparatus equipped with a 3% OV-225/AW-DMCS-Chromosorb W column (2.5 m  $\times$  3 mm) and an FID detector (detector temperature: 250 °C); column temperature programming: 110 °C (5 min)  $\rightarrow$ 210 °C (20 min).

Element analysis: VarioEL III elemental analyzer.

### Plant

The roots of *A. lappa* L. were the product of Gaomi City, Shandong Province, China, in March 2007 and identified by Ji'nan Botanical Garden and Professor Kao-Shan Chen in the School of Life Science, Shandong University (Ji'nan, China).

#### **Isolation of ALF**

The in shade air-dried roots (1 kg) were cut into pieces with a fodder chopper and extracted with EtOH (4 L) at room temperature for several days to remove fat and pigment. The residue was extracted successively with H<sub>2</sub>O (5 L, 2 h) at 80 °C for three times, filtered through gauze and centrifuged to remove water-insoluble materials. The aqueous extract was concentrated at 45 °C *in vacuo* and treated with 3 vols of 95% EtOH for precipitation at 4 °C overnight. The gel-like precipitate was solubilized in H<sub>2</sub>O and dialyzed against distilled H<sub>2</sub>O (exclusion limit 0.8 kDa). The nondialyzable portion was treated with papain-Sevag method to remove protein. After the process was repeated two times, the supernatant was lyophilized, and a yellow product was obtained named ALF1.

ALF1 was dissolved in distilled H<sub>2</sub>O, centrifuged and loaded onto a DEAE-cellulose column (80 cm  $\times$  3.5 cm). The column was eluted with H<sub>2</sub>O. Fractions of 10 mL were collected and monitored for the presence of carbohydrate using the phenol-H<sub>2</sub>SO<sub>4</sub> assay. Fractions containing carbohydrate were pooled, dialyzed and lyophilized. The fraction eluted with water from ALF1 designated as ALF2 was further fractionated on a column (100 cm  $\times$  3.5 cm) of Sephadex G-50, eluted with H<sub>2</sub>O. The carbohydrate was eluted as one single fraction (ALF) according to the elution profile. The fraction ALF was dialyzed and lyophilized.

#### Homogeneity and Molecular Weight

High-performance gel permeation chromatography (HPGPC) was carried out with a Waters 515 pump equipped

<sup>\*</sup>Address correspondence to this author at the College of Life Science, Qufu Normal University, Qufu, Shandong, 273165, China; Tel: +86 537 4456415; Fax: +86 537 4456887; E-mail: slsql2006@yahoo.com.cn

Table 1. GC-MS Data for Methylation Analysis of ALF

Methylated Sugars	Molar Ratio	Linkage Types
2, 3, 4, 6-Tetra- <i>O</i> -Me-Glc	1.0	$\operatorname{Glc}_p(1 \rightarrow$
1, 3, 4, 6-Tetra-O-Me-Fru	1.0	$Fru_{f}(2 \rightarrow$
3, 4, 6-Tri- <i>O</i> -Me-Fru	13	$\rightarrow 1$ ) Fru <sub>f</sub> (2 $\rightarrow$

with a Waters Ultrahydrogel<sup>TM</sup> 1000 column and a Waters 2410 RI detector. The column was calibrated with standard T-series Dextran (T-500, T-110, T-80, T-70, T-40 and T-9.3) with 0.003 M NaOAc as the mobile phase at a flow rate of 0.5 mL/min. All samples were prepared as 0.4% (w/v) solutions and 20  $\mu$ L of solution was analyzed in each run. The data were processed with Waters GPC Millennium<sup>32</sup> software.

#### Monosaccharide Analysis and Linkage Analysis [7]

ALF (3 mg) was hydrolyzed with 2 M TFA at 80 °C for 1 h, followed by evaporation to dryness. The residue was redissolved in H<sub>2</sub>O (0.2 mL), with 5  $\mu$ L of the solution used for TLC analysis. The other portion was successively reduced with NaBH<sub>4</sub>, acetylated with Ac<sub>2</sub>O at 100 °C for 1 h, and the resulting alditol acetate was examined by GLC.

ALF (5 mg) was methylated four times using the modified Ciucanu method [8]. The permethylated ALF was depolymerized with 90% formic acid (100 °C, 1 h), followed by hydrolysis with 2 M TFA (100 °C, 2 h). The hydrolysate was converted into partially methylated alditol acetate and analyzed by GC-MS with a Shimadzu QP Class-5000 instrument.

# <sup>1</sup>H NMR and <sup>13</sup>C NMR Spectroscopy

<sup>1</sup>H and <sup>13</sup>C NMR spectra were measured using a Bruker AM-400 NMR instrument equipped with a dual probe in the FT mode at 20 °C. ALF (30 mg) was dissolved in  $D_2O$  at a concentration of 30 mg/0.5 mL. Chemical shifts are referred

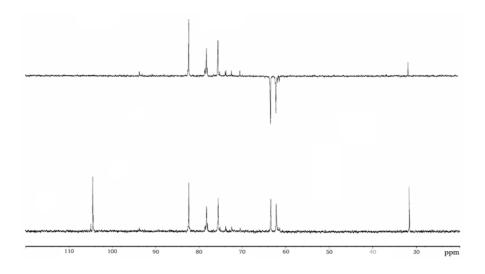
to the residual signal of HOD at  $\delta$  4.70 ppm for <sup>1</sup>H NMR spectrum and the internal standard, acetone for <sup>13</sup>C NMR spectra.

#### **RESULTS AND DISCUSSION**

ALF was obtained from the root of *A. lappa*, using water extraction, DEAE-cellulose column, and Sephadex G-50. On HPGPC, ALF showed a symmetrical peak, indicating a homogenous fraction. The average molecular weight was estimated to be  $3.1 \times 10^3$  g/mol. After complete hydrolysis with 2 M TFA, TLC analysis showed that ALF contains no uronic acid. GLC analysis indicated that it was composed of fructose and glucose. The absorption in the IR (not shown) indicated that ALF has characteristic peaks of polysaccharide. The content of C (40.53 %) and H (6.599 %) was estimated by elemental analysis. The molar ratio of C: H: O was 1:2:1. The [ $\alpha$ ]<sub>D</sub> value of ALF was -33 (*c* 0.80, H<sub>2</sub>O).

After methylating four times using the modified Ciucanu method, the O-H absorption at 3600-3200 cm<sup>-1</sup> in IR disappeared, indicating the completeness of methylation. The permethylated ALF was depolymerized and converted into partially methylated alditol acetate. GC-MS analysis showed three types of linkages, corresponding to T-Glc<sub>p</sub> (terminal-Glc<sub>p</sub>),  $\rightarrow$ 1) Fru<sub>f</sub> (2 $\rightarrow$  and Fru<sub>f</sub> (2 $\rightarrow$ , respectively, approximately in the molar ratio of 1:13:1 (Table 1).

The <sup>13</sup>C NMR spectrum contained two anomeric signals at  $\delta$  104.48 ppm and  $\delta$  93.58 ppm. The signal at  $\delta$  104.48 arose from the anomeric atoms of  $\rightarrow$ 1) Fru<sub>f</sub> (2 $\rightarrow$  and the signal at  $\delta$  93.58 was assigned to T-Glc<sub>p.</sub> The corresponding



	$\rightarrow$ 1) Fru <sub>f</sub> (2 $\rightarrow$	$\operatorname{Fru}_{f}(2 \rightarrow$	1-Glc <sub>p</sub>
C1	62.14	61.80	93.58
C2	104.48, 104.33	104.93	72.43
C3	78.53, 78.23	78.01	73.85
C4	75.52	75.01	70.84
C5	82.32 <sup>a</sup>	82.32 ª	73.65
C6	63.37 <sup>a</sup>	63.37 <sup>a</sup>	61.20

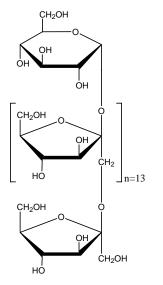
#### Table 2. <sup>13</sup>C NMR Signal Assignments for ALF

<sup>a</sup>Unresolved from other signals.

reversed peak in the DEPT spectrum, i.e.  $\delta$  63.37 and  $\delta$  62.14, were the signals of C6 and C1 of fructose (Fig. 1). Other <sup>13</sup>C NMR signals were tentatively assigned and are shown in Table **2**, referred to the literature values [9].

The structure of ALF was further confirmed by its HMQC and HMBC spectra (not shown). In the HMQC, the cross-peak was found showing a correlation between the C2 signal of the  $\beta$ -D-Fru<sub>f</sub> residue with the H-1 signal of the  $\alpha$ -D-Glc<sub>p</sub> residue, and there were no other cross-peaks between the signal of the  $\beta$ -D-Fru<sub>f</sub> residue with the signal of the  $\alpha$ -D-Glc<sub>p</sub> residue. Therefore, the  $\alpha$ -D-Glc<sub>p</sub> was shown to be linked only at the 1-position.

It could thus be concluded that ALF is comprised of  $\beta$ -D-Fru<sub>f</sub> and  $\alpha$ -D-Glc<sub>p</sub> in the molar ratio of 14:1, which is confirmed by the composition of 14 fructose residues linked by  $\beta$  (2 $\rightarrow$ 1) glycosidic bond and 1 glucose residue linked by  $\alpha$ (1 $\rightarrow$ 2) glycosidic bond at the end of linear straight sugar chain. The idealized repeating unit of the fructan isolated from roots of *Arctium lappa* L. (i.e. the structure formula of ALF) is proposed to be:



It is the first time that the presence of the fructan possessing the DB value in *A. lappa* is reported.

#### ACKNOWLEDGEMENTS

The structure determination of this novel fructan was performed at the glycochemistry & glycobiology lab in Shanghai Institute of Materia Medica, Chinese Academy of Sciences, China led by Professor Ding Kan. Great thanks were expressed to Professor Ding Kan and his lab. This study was supported by the Research Fund of Qufu Normal University (No. XJ200822).

#### REFERENCES

- Andrew JC, Paul B, Ian MS. The structure of starch from seeds and leaves of the fructan-accumulating ryegrass, *Lolium temulentum* L. J Plant Physiol 2002; 159: 221-30.
- [2] Praznik W, Spies T. Fructo-oligosaccharides from Urginea maritime. Carbohydr Res 1993; 243: 91-7.
- [3] Hincha DK, Zuther E, Hellwege EM, Heyer AG. Specific effects of fructo- and gluco-oligosaccharides in the preservation of liposomes during drying. Glycobiology 2002; 12: 103-10.
- [4] Kato Y, Watanabe T. Isolation and characterization of a xyloglucan from gobo (*Arctium lappa L.*). Biosci Biotechnol Biochem 1993; 57: 1591-2.
- [5] Healy ML, Rogan CJ, Fekete FA, Mundy BP. Book of Abstracts: 1999: Proceedings of 217th ACS National Meeting; Mar 21-25 1999; Anaheim, USA 1999.
- [6] Kardosová A, Ebringerová A, Alföldi J, Nosál'ová G, Franová S, Hríbalová V. A biologically active fructan from the roots of Arctium lappa L., var. Herkules. Int J Biol Macromol 2003; 33: 135-40
- [7] Dong Q, Fang JN. Structural elucidation of a new arabinogalactan from the leaves of *Nerium indicum*. Carbohydr Res 2001; 332: 109-14.

[8] Shi L, Fu YL, Chen KS. A novel water-soluble α-(1→4)-glucan from the root of *Cudrania tricuspidata*. Fitoterapia 2007; 78: 298-301.

[9] Chen XM, Tian GY. Structural elucidation and antitumor activity of a fructan from *Cyathula officinalis* Kuan. Carbohydr Res 2003; 338: 1235-41.

Received: April 14, 2009

Revised: June 08, 2009

Accepted: June 12, 2009

© Shi Lei; Licensee Bentham Open.

This is an open access article licensed under the terms of the Creative Commons Attribution Non-Commercial License (http://creativecommons.org/licenses/by-nc/3.0/) which permits unrestricted, non-commercial use, distribution and reproduction in any medium, provided the work is properly cited.